

International Journal of Ayurveda and Pharma Research

Research Article

PHARMACEUTICO- ANALYTICAL STUDY OF RASAMANIKYA PREPARED BY VARIOUS METHODS

Santosh Kulkarni

Professor, Department of Rasashastra & BK, S J G Ayurvedic Medical College, Koppal, Karnataka, India.

ABSTRACT

Rasamanikya prepared from *Shodhita Haritala* is one of the effective and economical medicines used in different skin and respiratory disorders. Different methods are explained in the classics for the preparation of *Rasamanikya* and few adopted methods are also developed from scholars of *Rasashastra* depending upon their experience. Present study is aimed at exploring all these methods in detail and any modifications if needed. The *Patra Haritala* is subjected to different *Shodhana* procedures in different media, the changes observed are discussed in the article. The yield of *Shodhita Haritala* was 92% to 96% in different *Shodhana* methods.

Rasmanikya is prepared in six different methods. *Sharava samputa* method (Method III & IV) can be considered as suitable methods for pharmaceutical preparation of *Rasamanikya* in large quantity as there was loss of only 11% to 13% drug was noticed. Chemical analysis and X-Ray diffraction of *Patra Haritala* and *Rasamanikya* prepared from all the methods is carried as a part of standardization. In Analytical study of all the methods, 44% to 47% of Arsenic and 22 to 29% of Sulphur was present in *Rasamanikya*. X-RD study of *Haritala* and *Rasamanikya* samples revealed that crystalline form of *Haritala* was changed to relatively amorphous form in *Rasamanikya* prepared by I, II, III and IV methods, which indicates quick and better absorption of the drug *Rasamanikya* on administration making it one of the economical and potent medicine. Chemically *Rasamanikya* can be considered as a complex compound of As₂S₃ and As₂O₅.

KEYWORDS: Rasamanikya, *Sharava samputa*, X-RD study, As₂S₃ and As₂O_{5.}

INTRODUCTION

Haritala is a well-known drug since Vedic period. It has a unique place in *Dehavada* and *Dhatuvada*. *Rasamanikya* is a simple unique medicine as it only contains *Patra Haritala*. It is prepared by heating *Patra Haritala* till it attains *Manikya varna*. In present study *Rasamanikya* is prepared by 6 different methods and subjected to analytical study including X-RD to explore the chemical nature of *Rasamanikya*. The details of the study are as follows:

Patra Haritala (As₂S₃) was obtained from SDM Pharmacy, Udupi. It is observed for all *Grahya lakshana* mentioned in *Rasashastra* classics.^[1]

IJAP

The *Patra Haritala* is subjected to *Shodhana* and *Rasmanikya* (RM) prepared by 6 methods. The details are as follows:

RM1- Glass bottle method (Adapted method)

RM2 - *Abhraka Patra* method (R T 11/90-93)

RM3- Modified *Sharava Samputa* method (R C 9/128-131)

RM4-Modified Sharava Samputa method (RS S1/182)

RM5 - Churnodaka method (Adapted)

RM6- Kushmanda Swarasa Swedana method

Hartala Shodhana	Rasa manikya preparation				
i) <i>Patra Haritala is</i> (230gm) subjected to <i>Swedana</i> in	i) <i>Shodhita Haritala</i> was grinded in mortar and pestle and taken ir small glass bottle filled up to $3/4$ th				
Churnodaka ^[2]	ii) Glass bottles were placed over a wire plate which is kept on gas				
ii) Churnodaka Swedita Patraharitala	stove and subjected to moderate fire for 10 to 20 minutes.				
was kept in <i>Kushmandaswarasa</i> (2 ltrs) and <i>Amladadhi</i> (2 ltrs) for 3 days	iii) When colour of <i>Haritala</i> completely changes to Ruby red, a sticl is introduced into the middle of the bottle to assess <i>Tantupaka</i> .				
respectively.	iv) Glass bottle taken out from the gas stove and were rolled in cold				

Table 1: Rasa Manikya prepared by 1st Method (RM1)

Santosh Kulkarni. Pharmaceutico- Analytical Study of Rasamanikya Prepared by Various Methods

-	water by which glass bottles were broken.
water and weighed after complete	v) Rasa Manikya collected carefully and grinded into fine powder
drying (195gm). (Photo plate I)	and filtered through the cloth. (Photo plate II)

Table 2: Rasa Manikva	prepared by 2 nd Method (RM2)

Hartala Shodhana	Rasa manikya preparation (R T 11/92-93)
i) <i>Patrahartala</i> was grinded in <i>Khalva Yantra</i> and subjected to <i>Bhavana</i> with <i>Kushmanda</i> <i>Swarasa</i> and <i>Amladadhi</i> for 3 times separately and dried well ^[3] (R.T. 11/90) (Photo plate I)	ng Burning Gharcour was tanten in a Sharava and Shirota abin ana patra were

	Table 3: <i>Rasa Manikya</i> prepared by 3 rd Method (RM3)									
Hartala Shodhana	Rasa manikya preparation (R C9 /128-131)									
i) <i>Patra Harital</i> was purified by keeping	i) <i>Shodhita Haritala</i> spread evenly in a <i>Sharava</i> and covered with another <i>Sharava</i> which had hole in the middle. ^[5]									
it in <i>Kushmanda</i> <i>swarasa</i> & <i>Amla</i> <i>dadhi</i> for 3days in each separately. (Photo plate I)	ii) Sharava Samputa is done and Sharava is kept on moderate fire.									
	iii) The hole of the upper Sharava was initially closed with plug of Multani Mitti.									
	iv) When fumes observed the plug of <i>Multani Mitti</i> was removed from the upper <i>Sharava</i> .									
	v) A stick was passed through the hole to observe the <i>Tantupak</i> and <i>Sharavsamputa</i> taken out from the fire.									
	vi) After cooling <i>Rasamanikya</i> was collected from lower <i>Sharava</i> then it is grinded into fine power and filtered out through the cloth. (Photo plate III)									
	Table 4: <i>Rasa Manikya</i> prepared by 4 th Method (RM4)									

Hartala Shodhana	Rasa Manikya Preparation
i) Patraharitala is purified by Swedana in Dolayantra containing Kushmanda Swarasa for 3 hours and Amladadi for 3 hours one after the other. ^[6] (RSS 1/182) (Photo plate I)	 i) Shodhita Haritala was crushed into small size and spread evenly in a Sharava which is covered with another Sharava having hole in the middle. Sharava Samput was done and it is subjected to moderate fire. ii) The hole of the upper Sharava was initially closed with plug of Multani Mitti. iii) When fumes observed the plug of Multani Mitti was removed from the upper Sharava. iv) A stick was passed through the hole to observe the Tantupaka and Sharavsamputa taken out from the fire. v) After cooling Rasamanikya was collected from lower Sharava and filtered out through the cloth.
Table	e 5: <i>Rasa Manikya</i> prepared by Glass Bottle Method (RM5)

Hartala Shodhana	Rasa manikya preparation
i) Patraharitala is purified by Swedana in Dolayantra containing Churnodaka for 3 hours. (Photo plate I)	 i) Shodhita Patra Haritra was taken in small, clean glass bottles filled up to 3/4th. ii) A wire plate was kept on gas stove; 2-3 bottles were placed on it. iii) It was subjected to moderate fire when the glass bottles turned to red; they were kept transversely on the wire plate. iv) When the colour of Haritala changed into ruby red, a stick was introduced into the middle and observed for Tantupak and taken out from fire. v) The hot glass bottles rolled in cold water, due to sudden cooling glass bottles broken down and Rasamanikya is collected carefully. vi) Rasamanikya is grinded into fine powder and filtered through the cloth.

Int. J. Ayur. Pharma Research, 2019;7(5):38-44

Tab	le 6: Rasa Manikya prepared by Glass Bottle Method (RM6)
Hartala Shodhana	Rasa manikya preparation
i) Patraharitala is purified by Swedana in Dolayantra containing Kushmanda Swarasa for 3 hours. (Photo plate 1)	were kept transversely on the wire plate.

S.No.	Sample	Colour	Taste	Touch	Smell	Yield%	Time	Temp ⁰ C				
1	RM 1	Istika Varna	Tasteless	Fine	Odor less	93.88	12min	180				
2	RM 2	Black	Tasteless	Fine	Odor less	93	4min	190				
3	RM3	Pale brick	Tasteless	Fine	Odor less	88.88	45min	200				
4	RM4	Brownish yellow	Tasteless	Fine	Odor less	88.26	42min	200				
5	RM5	Red	Tasteless	Fine	Odor less	93.3	12min	180				
6	RM6	Brown	Tasteless	Fine	Odor less	91.11	13min	180				

Table 7: Observations noted during above six methods. (Photo plate IV)

Analytical study

The samples of *Rasamanikya* prepared by above six methods collected separately and subjected to analytical study through organoleptic, qualitative and quantitative analysis.

Organoleptic study

Table 8: Organoleptic study of RM by different methods

S.No	Sample	Colour	Taste 5	Touch	Smell
1	RM 1	Istika Va <mark>rna</mark>	Tasteless	Fine	Odor less
2	RM 2	Black	Tasteless	Fine	Odor less
3	RM3	Pale brick	Tasteless	Fine	Odor less
4	RM4	Brownish yellow	Tasteless	Fine	Odor less
5	RM5	Red	Tasteless	Fine	Odor less
6	RM6	Brown	Tasteless	Fine	Odor less

Qualitative study

Nambudari phase spot test

0.25gm fine powder of RM was dissolved in 0.5ml of concentrated HNO₃, 5N HNO₃ and Aquaregia in 3 separate test tubes heated and kept for 24hrs. Next day a drop of solution from each tube was put on chemically reacting 10% KI paper and 5% Potassium Ferrocynide paper and observed for the development of spots in three phases.

In the third phase, a central white coloured spot observed with con HNO_3 on 10% KI paper. Central green coloured spots were observed on 5% potassium ferrocynide paper by all solutions.

Quantitative study

Rasamanikya samples of each method collected separately and subjected quantitative analysis as follows. Glass bottle method (I method)

S1= Raw drug *Patra Haritala*

S2= Powder of Ashuddha Haritala subjected to Swedana in Churnodaka

- S3= *Haritala* tied in a *Pottali* kept in *Kushmanda Swarasa* for 3 days
- S4= Haritala tied in a Pottali kept in Amladadhi for 3 days

S5= Rasamanikya

	Tuble 91 Quantitative intarysis of Habamanny'a by Thechou in 70												
Sample	As	S	Fe	Mg	Ca	Silica	Moisture	Ash	Water	Acid	Water	PH	
								value	Insoluble	Insoluble	Insoluble		
										Ash	Ash		
S1	44.031	20.90	0.11	0.27	0.43	0.099	0.191	0.88	86.66	0.099	1.69	6.78	
S2	46.069	24.90	0.11	0.24	0.43	0.159	-	0.47	83.66	0.15	1.19	7.28	
S3	45.84	25.50	0.22	0.02	0.45	0.119	0.009	-	87.18	0.019	0.88	6.63	
S4	47.141	27.68	0.11	0.02	0.46	0.159	-	0.5	84.03	0.15	2.30	7.76	
S5	47.058	28.58	0.11	0.02	0.54	0.2	0.2	0.54	86.40	0.2	1.7	7.78	

Table 9: Quantitative Analysis of Rasamanikya by I Method in %

Abhraka Patra method (II method)

S1= Raw drug Patra Haritala

S2= Patra Haritala after giving 3 Bhavana of Kushmanda Swarasa

S3= Patra Haritala after giving 3 Bhavana of Amla Dadhi

S4= Rasamanikya

Table 10: Quantitative Analysis of Rasamanikya by II Method in %

Sample	As	S	Fe	Mg	Ca	Silica	Moisture	Ash	Water	Acid	Water	РН
								value	Insoluble	Insoluble	Insoluble	
											Ash	
S1	44.031	20.90	0.11	0.027	0.43	0.099	0.1912	0.88	86.66	0.099	1.69	6.78
S2	46.179	20.98	0.86	0.63	0.93	0.69	0.0976	-	82.95	0.69	3.25	6.4
S3	46.26	21.99	087	0.64	0.78	0.31	0.0097	-	84.27	0.31	1.64	4.83
S4	47.24	22.98	1.04	0.89	0.80	0.65	0.0991	-	98.60	0.65	1.64	6.68

Modified Sharava method (III method)

S1= Raw drug Patra Haritala

S2= Patra Haritala after keeping 3 days in Kushmanda Swarasa

S3= Patra Haritala after keeping 3 days in Amla Dadhi

S4= Rasamanikya

Table 11: Quantitative Analysis of *Rasamanikya* by III Method in %

Sample	As	S	Fe	Mg	Ca	Silica	Moisture	Ash	Water	Acid	Water	РН
								value	Insoluble Insoluble		Insoluble	
											Ash	
S1	44.031	20.90	0.11	0.027	0.43	0.099	0.1912	0.88	86.66	0.099	1.69	6.78
S2	44.05	24.17	0.50	0.49	0.78	0.72	0.19	-	79.4	0.72	2.51	6.71
S3	47.22	24.59	0.45	0.11	0.70	0.32	0.25	-	93.05	0.32	2.31	6.29
S4	44.80	23.32	0.43	0.42	0.78	0.38	0.29	-	96.62	0.38	3.38	5.97

Modified Sharava method (IV method)

S1= Raw drug Patra Haritala

S2= Patra Haritala subjected for Swedana in Kushmanda Swarasa for 3hrs.

S3= Patra Haritala subjected for Swedana in Amla Dadhi for 3hrs.

S4= Rasamanikya

Table 12: Quantitative Analysis of *Rasamanikya* by IV Method in %

Sample	As	S	Fe	Mg	Са	Silica	Moisture	Ash	Water	Acid	Water	PH
								value	Insoluble	Insoluble	Insoluble	
											Ash	
S1	44.031	20.90	0.11	0.027	0.43	0.099	0.1912	0.88	86.66	0.099	1.69	6.78
S2	44.35	22.17	0.44	0.10	0.67	0.39	0.99	-	90.64	0.39	0.38	6.18
S3	44.50	23.35	0.44	0.10	0.68	0.31	0.009	-	84.62	0.31	0.37	6.03
S4	44.52	28.87	0.44	0.10	0.67	0.55	0.09	-	85.60	0.55	0.39	6.02

Churnodaka method (V method)

S1= Raw drug *Patra Haritala*

S2= Patra Haritala subjected to Swedana in Churnodaka for 3 hrs.

S3= Rasamanikva

Table 13: Quantitative Analysis of *Rasamanikva* by V Method in %

Sample	As	S	Fe	Mg	Ca	Silica	Moisture	Ash	Water	Acid	Water	PH
								value	Insoluble	Insoluble	Insoluble Ash	
S1	44.031	20.90	0.11	0.027	043	0 099	0.1912	0.88	86.66	0.099	1.69	6.78
S1 S2	46.11	24.9	0.11	0.027	0.43		-	0.47	83.66	0.15	1.19	7.28
S3	46.54	26.18	0.45	0.11	0.81	0.64	0.29	-	94.58	0.56	0.38	6.82

Kushmanda Swarasa Swedana method (VI method)

S1= Raw drug *Partra Haritala*

S2= Patra Haritala after Swedana in Kushmanda Swarasa

S3= Rasamanikya

Table 14: Quantitative Analysis of Rasamanikva by VI Method in %

Sample	As	S	Fe	Mg	Са	Silica	Moisture	Ash value	Water Insoluble	Acid Insoluble	Water Insoluble Ash	РН
S1	44.031	20.90	0.11	0.027	0.43	0.099	0.1912	0.88	86.66	0.099	1.69	6.78
S2	44.37	22.17	0.44	0.10	0.95	0.39	0.99	-	90.64	0.39	0.38	6.18
S3	47.39	29.73	0.92	0.22	0.90	0.57	0.49	-	94.18	0.57	0.27	6.85

Particle size determination

Microscopic examination of samples for particle size determination with filar eve piece revealed size as follows. Table 15: Particle size determination

		lable	15:	Partic	cie s	size	aeter	mir	lation				
Particle	Particle S Size 74.4µm		L Jo	RM2		RM	13 🚺	RM4		R	M5	RM6	
Size			um 🛛 🛛 🛛		n	90 µm		60 µm		204 µm		60 µm	
	Table 16: Surface area determination												
Surface	Surface S				RM2		RM3		RM4		RM5	RM6	
Area (Sc	l.m/gm)	1.11	0.28		0.4	12	0.14		0.22		0.28	0.34	
	Table 17: Sieve analysis												
S. No	Sieve	no	Retained				% Retained				% Passed		
1	170B	SS	16.960Gms				33.920				66.080		
2	200 B	SS	5.044Gms				10.088				89.912		
3	240B	5.237Gms				10.474			8	89.526			
4	300 B	300 BSS			12.99Gms			2.598			97.402		
5	350BSS		0.202Gms				0.404			Ç	99.596		
6	6 Pan			110Gm		12.220							

X- Ray Diffraction study

Based on the peak developed by each sample during X-RD study following interpretation can be made.

Haritala sample

- 1. The sample is more crystalline in nature
- 2. It showed prominent peak of As₂S_{3.}

RM 1 sample

- 1. It is relatively amorphous
- 2. There is a prominent peak of As_2S_3 and As_2O_5 and also some peaks of AsS

RM 2 Sample

1. It is relatively amorphous in nature.

2. There is a peak mainly contributed by As_2O_5 and $Cu_5(AsO_4)_2(0H)_2$

RM 3 Sample

- 1. It is relatively amorphous in nature
- 2. There is a prominent peak contributed by As_2S_3 and As₂₀₅
- 3. Small peaks indicate little quantity of AsS and very little quantity of As₂o₃

RM 4 Sample

- 1. It is relatively amorphous in nature
- 2. The prominent peak is contributed by As_2S_3 and

 $Cu_5 (AsO_4)_2 (OH)_4$

3. It also contains little quantity of As_2O_3

RM 5 Sample

- 1. It is relatively crystalline in nature
- 2. The shift of the higher peak compared to other sample is due to the formation of new compound Cu_5H_2 (AsO₄)4H₂O Along with As₂S₃

RM 6 Sample

1. It is relatively crystalline in nature

2. As_2S_3 and As_2O_5 contribute to the prominent peak **DISCUSSION**

Haritala shodhana

As Haritala is considered as Dhatu visha^[7], different media were used for Shodhana. In the first method of Rasamanikya preparation Haritala was subjected to Shodhana in Churnodaka^[2] which is a basic media hence it neutralizes acidic impurities of Haritala. Then Haritala was kept in Kushmanda Rasa^[3] and Amladadhi^[3] which are acidic media, neutralizing the basic impurities of Haritala. This rationality can be applied for the use of different media in Haaritala Shodhana.

There was no much difference in the yield of *Haritala* in different *Shodhana* procedures. 92% to 96% yield was observed.

Pharmaceutical study

The method of preparation of *Rasamanikya* can be divided in to two types.

- a) **Closed method:** A_{s2}S₃ heated in limited supply of oxygen, loss of sulphur as oxides are less e.g., *Sharava Samputa* method, *Abhraka Patra* method.
- b) **Open method:** In this method oxygen supply is more, loss of sulphur as oxides is more. E.g; Glass bottle method.

But by analytical study there was no much difference observed in the quantity of Sulphur, hence there was no much difference in yield of *Rasamanikya* was noted in between these two methods.

Analytical study

From the quantitative analysis of all the six methods Arsenic and Sulphur were found to be main contents of the drug. During *Haritala Shodhana* and *Rasa Manikya* preparation in each method it is observed that percentage of sulphur and Arsenic were increased. It can be justified as the substances used in *Shodhana* are enriched with organic compounds containing elements of carbon and hydrogen. During *Shodhana* these elements may make some linkage with the elements present in crude As_2S_3 causing an increase in the percentage of Arsenic after *Shodhana*.

During the practicals it was observed that weight of RM reduced as compared to *Haritala* sample. It may be due to evaporation of volatile contents during the procedure. As the weight decreases concentration of arsenic and Sulphur increased in RM.

X-RD study: In most of the methods sulphide form of *Haritala* was changed to oxide form and a complex compound of As_2S_3 and As_2O_5 was observed in prominent peaks.

Crystalline form of *Haritala* was changed to amorphous form in RM 1,2,3,4 samples, which indicates better absorption of the drug in the body.

CONCLUSION

The lustre and smell of *Patra Haritala* were decreased after different methods of *Shodhana*. Along with *Manikya Varna*, the *Tantu Paka* should also be considered as criteria for the completion of *Rasamanikya* preparation. *Rasamanikya* powder by different methods showed different colours, but analytically there is no marked difference between the *Rasamanikya* prepared by different methods.

The *Sharava Samputa* method with suitable modifications is suitable for the large-scale pharmaceutical preparation of *Rasamanikya*.



By X-RD study chemically *Rasamanikya* can be considered as a complex compound of As₂S₃ and As₂O_{5.}



ACKNOWLEDGEMENT

This work is based on my PG Dissertation, my sincere thanks to my PG guide Prof Rajashekhara Pandey for his guidance throughout the work with his valuable suggestions.

I am thankful to Dr Sujata k, Professor, SDM college of Ayurveda, Bangalore for her help & suggestions throughout the work.

REFERENCES

- 1. Vagbhatacharya, Vignana Bhodhini Tika, Prof D.A Kulkarni, Rasaratnana samuchhaya, New Dehli, Mehachanda Lachhaman das publications,1998, 3rd chapter, 66th Shloka, pg no 53.
- Vagbhatacharya, Vignana bhodhini tika, Prof D.A Kulkarni, Rasaratnana samuchhaya, New Dehli, Mehachanda Lachhaman das publications,1998, 3rd chapter, 70th Shloka, pg no 54.
- 3. Sadanandasharma, Haridatta shastri, Rasa

Cite this article as:

Santosh Kulkarni. Pharmaceutico- Analytical Study of Rasamanikya Prepared by Various Methods. International Journal of Ayurveda and Pharma Research. 2019;7(5):38-44.

Source of support: Nil, Conflict of interest: None Declared

tarangini, New Dehli, Motilala Banarasi das, 2004, 11th Taranga, 90th shloka, pg no258.

- 4. Sadanandasharma, Haridatta shastri, Rasatarangini, New Dehli, Motilala Banarasi das, 2004, 11th Taranga, 92nd shloka, pg no258.
- Dundukanath, Siddhiprada tika, Prof Siddinandan Mishra, Rasendra Chintamani, 1st edition, Varanasi, Chaukhambha Orientalia, 2000, 9th chapter, 128th to 133rd shloka, pg no 376.
- Sri Gopal Krishna, Satyartha Prakash, Rasendra sara sangraha,1st edition Varanasi, Krishnadas academy, 1994, 1st chapter, 182 to184 shloka, pg no 121.
- 7. Sushruta, Nibhanda sangraha Commentary of Dalhana, Edited by Yadavaji Trikramaji Acharya, 8th edition, Varanasi, 2005, pg no 564.

*Address for correspondence Dr Santosh Kulkarni Professor, Department of Rasashastra & BK, S J G Ayurvedic Medical College, Koppal, Karnataka, India. Email: <u>drsantoshsk64@gmail.com</u> Mobile: 9916288585

Disclaimer: IJAPR is solely owned by Mahadev Publications - dedicated to publish quality research, while every effort has been taken to verify the accuracy of the content published in our Journal. IJAPR cannot accept any responsibility or liability for the articles content which are published. The views expressed in articles by our contributing authors are not necessarily those of IJAPR editor or editorial board members.